

STUDY MODULE DESCRIPTION FORM		
Name of the module/subject Heat and Mass Transfer		Code 1010135211010130346
Field of study Enviromental Engineering Extramural Second-	Profile of study (general academic, practical) (brak)	Year /Semester 1 / 1
Elective path/specialty Heating, Air Conditioning and And	Subject offered in: Polish	Course (compulsory, elective) obligatory
Cycle of study: Second-cycle studies	Form of study (full-time,part-time) part-time	
No. of hours Lecture: 20 Classes: - Laboratory: 10 Project/seminars: -		No. of credits 4
Status of the course in the study program (Basic, major, other) (brak)		(university-wide, from another field) (brak)
Education areas and fields of science and art technical sciences		ECTS distribution (number and %) 4 100%
Responsible for subject / lecturer: Prof. dr hab. inż. Janusz Wojtkowiak/Prof. dr hab. inż. Czesław Oleśkiewicz-Popiel email: czeslaw.oleskowicz-popiel@put.poznan.pl tel. 061 6652-537 Faculty of Civil and Environmental Engineering ul. Berdychowo 4, 61-131 Poznań		Responsible for subject / lecturer: Prof. dr hab. inż. Janusz WOJTKOWIAK email: janusz.wojtkowiak@put.poznan.pl tel. 61 665-2442 Faculty of Civil and Environmental Engineering ul. Berdychowo 4, 61-131 Poznań
Prerequisites in terms of knowledge, skills and social competencies:		
1	Knowledge	Mathematics: differential and integral equations and their solutions. Physics at level 5KRK. Thermodynamics at level 6KRK. Fluid mechanics at level 6 KRK.
2	Skills	Application of differential and integral equations for description of physical phenomena, solution of differential equations. Thermodynamics: analysis of thermodynamic problems and realization of measurements and investigations at level 6 KRK
3	Social competencies	Awareness of the need of permanent updating and supplementing knowledge and engineering skills.
Assumptions and objectives of the course: Extension of knowledge and skills of heat and mass transfer calculations and measurement of heat fluxes appearing in build and natural environmental engineering.		
Study outcomes and reference to the educational results for a field of study		
Knowledge:		
1. Student has a wider knowledge and calculation skill concerning heat and mass transfer and heat measurements needed for solving advanced thermal problems in environmental engineering appearing in build and natural environment. (achieved during lectures and tutorials) - [K2_W03, K2_W04, K2_W07]		
2. Student knows basic methods needed for theoretical and design solution of thermal problems appearing in environmental engineering. (achieved during lectures and tutorials) - [K2_W03, K2_W04, K2_W07]		
3. Student has a knowledge concerning detailed rules of methods and calculations of heat and mass transfer and heat exchangers appearing in environmental engineering. (achieved during lectures, tutorials and laboratory exercises) - [K2_W03, K2_W04, K2_W07]		
4. Student has an advanced knowledge concerning development tendencies and new achievements in heat equipment and processes in environmental engineering. (achieved during lectures, tutorials and laboratory exercises) - [K2_W03, K2_W04, K2_W07]		
Skills:		

1. Student can find and estimate information gained from literature and internet concerning heat engineering equipment appearing in environmental engineering. (achieved during lectures, tutorials and laboratory exercises) - [K2_U01, K2_U18]
2. Student can find adequate relationships describing analysed heat processes. (achieved during lectures and tutorials) - [K2_U01, K2_U18]
3. Student knows how determine thermodynamic properties needed for heat and mass transfer calculations (achieved during tutorials and laboratory exercises) - [K2_U01, K2_U18]
4. Student can recognize and solve advanced design and operation heat and mass transfer problems occurred in heat equipment (achieved during tutorials and laboratory exercises) - [K2_U01, K2_U18]
5. Student can critically estimate a design solution and recognize a danger hazard in erected and operated heat equipment (achieved during lectures, tutorials and laboratory exercises) - [K2_U01, K2_U18]
6. Student can plan and realize operating tests and prototype investigations of equipment appearing in environmental engineering (achieved during tutorials and laboratory exercises) - [K2_U01, K2_U18]
7. Student can determine an accuracy and analyse obtained results of calculation and measurements (achieved during tutorials and laboratory exercises) - [K2_U01, K2_U18]
8. Student can critically analyse obtained results of calculations and measurements and develop conclusions (achieved during lectures, tutorials and laboratory exercises) - [K2_U01, K2_U18]

Social competencies:

1. Student understand and appreciate a team cooperation in solving theoretical and practical problems (achieved during lectures, tutorials and laboratory exercises) - [K2_K03]
2. Student is aware of the range and limits of the used calculation methods and relationships as well as importance of possessed a theoretical and practical knowledge (achieved during lectures, tutorials and laboratory exercises) - [K2_K03]
3. Student understands a need of examination and verification of results of used calculation and experimental methods.(achieved during lectures, tutorials and laboratory exercises) - [K2_K03]
4. Student is aware of need of innovation (achieved during lectures, tutorials and laboratory exercises) - [K2_K03]

Assessment methods of study outcomes

Lecture (results W03, W04, W07, U01, U18)

The final exam consists of two parts:

Part 1: Test of understanding of fundamentals of heat and mass transfer (3 to 5 questions).

Part 2: Test of competence conc. solving of heat and mass transfer engineering problems (1 to 3 problems).

In some cases an oral examination is used. Also an activity of students during lectures and tutorials is taken into account.

To pass each of the two parts of the exam (as well as to pass the tutorials) there is necessary to obtain at least 50% of the maximum points (max=20 points). The exam is passed if both part 1 and part 2 are passed. Corrected (Improved) is only this part which was failed.

Grading system:

0-9 points = 2,0 (failed)

10-12 points = 3,0 (sufficient)

13-14 points = 3,5 (sufficient plus)

15-16 points = 4,0 (good)

17-18 points = 4,5 (good plus)

19-20 points = 5,0 (very good)

Tutorials (results U01, U18)

60-minute written test at the end of semester (solution of of 1 or two problems). Continuous assessment of student activity (reward of activity)

Laboratory training (exercise):

Short 15 minute entrance test before each laboratory training and final assessment of the written raport and eventual oral presentation of the results.

Course description

Introduction to heat transfer. Heat flux. Mechanisms of heat transfer by conduction, convection and radiation. Fourier law, heat conduction equation, boundary and initial conditions. Conductivity and overall heat transfer through flat plate, cylindrical walls and finned walls. Theory of single fin. Two-dimensional steady-state conduction. Transient heat conduction lumped capacitance method, Biot and Fourier numbers. Heating and cooling of plate and regular bodies. Introduction to the numerical methods. Introduction to convection, continuity, momentum and energy equation, boundary layer equations, Nusselt, Reynolds, Prandtl and Grashof numbers. Heat convection in external and internal flows. Free convection. Convection in atmospheric air. Convection in boiling and condensation. Heat transfer by radiation, solar radiation.

Heat exchangers, types of heat exchangers, theory of heat exchanger, overall heat transfer coefficient, log mean temperature difference. Methods of heat exchanger calculations: logarithmic and effectiveness-NTU methods. Compact heat exchangers. Effect of fouling on heat exchanger rate and pressure losses.

Diffusion mass transfer, Fick law, diffusion coefficient. Diffusion in gases, liquids and solids. Introduction to mass convection: Sherwood and Schmidt numbers. Correlation equations for mass convection. Momentum, heat and mass transfer analogy. Application of mass and heat convection for psychrometry (wet and dry thermometers), evaporation of liquid water in atmospheric air.

Teaching method:

Classical lecture with elements of conversation

Tutorials ? solving problems method

Laboratory exercises ? teaching by experimentation.

Basic bibliography:

1. Wiśniewski St., Wiśniewski T.S., Wymiana ciepła. WNT, Warszawa 2000
2. Kostowski E., Przepływ ciepła. Wyd. Polit. Śląskiej, Gliwice, 1986
3. Madejski J., Teoria wymiany ciepła. Wyd. Ucz. Politechniki Szczecińskiej, Szczecin 1998
4. Oleśkiewicz-Popiel C., Wojtkowiak J., Eksperymenty w wymianie ciepła. Wyd. II rozszerzone, Wyd. Politechniki Poznańskiej, Poznań 2007
5. Oleśkiewicz-Popiel C., Wojtkowiak J., Właściwości termofizyczne powietrza i wody-PRZEZNACZONE DO OBLICZEŃ PRZEPŁYWÓW I WYMIANY CIEPŁA. Wyd. Politechniki Poznańskiej, Poznań 2010
6. Staniszewski B., Wymiana ciepła. Podstawy teoretyczne. PWN, Warszawa 1979, 1980
7. Pomiary cieplne, T. 1 i T. 2, Praca zb. (red. T.R. Fodemski), WNT, Warszawa 2001
8. Hobler T., Ruch ciepła i wymienniki. WNT, Warszawa 1979
9. Kalinowski E., Przekazywanie ciepła i wymienniki. Skr. Pol. Wr., Wrocław 1995
10. Zbiór zadań z przepływu ciepła, pod red. E. Kostowskiego, Wyd. Polit. Śląskiej, Gliwice 2001
11. Oleśkiewicz-Popiel C., Czujniki strumieni ciepła. Wyd. Politechniki Poznańskiej, Poznań 1986
12. Pogorzelski J.A., Fizyka cieplna budowli, PWN, 1976
13. Modelowanie numeryczne pól temperatury. Pod red. J. Szarguta. WNT, Warszawa 1992
14. Taler J., Duda P., Rozwiązywanie prostych i odwrotnych zagadnień przewodzenia ciepła. WNY, Warszawa, 2003
15. Zarzycki R., Wymiana ciepła i ruch masy w inżynierii środowiska. WNT, Warszawa 2005

Additional bibliography:

1. Incropera F.P., De Witt D.P., Bergman T.L., Lavine A.S., Introduction to Heat and Mass Transfer. 5th Ed., John Wiley and Sons, 2007
2. Incropera F.P., De Witt D.P., Bergman T.L., Lavine A.S., Fundamentals of Heat and Mass Transfer. 6th Ed., John Wiley and Sons, 2006
3. Bejan A., Kraus A.D., Heat Transfer Handbook, John Wiley & Sons, Inc., 2003
4. Eckert E.R.G., Drake R.M., Analysis of Heat and Mass Transfer. McGraw-Hill Book Co., 1972
5. Holman J.P., Heat Transfer, McGraw Hill, Metric Edition, 1989
6. Kakac S., Liu H., Heat exchangers: Selection, Rating, and Design. CRC Press, 1998
7. Howell J.R., Siegel R., Menguc M.P., Thermal Radiation Heat Transfer. CRC Press and Taylor and Francis Group, New York 2011

Result of average student's workload

Activity	Time (working hours)
1. Lectures (contact hours)	20
2. Participation in laboratory training (exercises) (contact hours, practical training)	10
3. Preparation to laboratory exercises (autonomous learning)	20
4. Development of laboratory reports and oral presentation (autonomous learning)	12
5. Consultations (contact hours)	3
6. Preparation to examination test and presence at the exam (autonomous learning and contact hours)	35

Student's workload

Source of workload	hours	ECTS
Total workload	100	4
Contact hours	35	1
Practical activities	10	0